

Laboratory Level Rubric	Pre-Laboratory			Laboratory	Post-Laboratory	
<b>Student &amp; Teacher Responsibility</b>	<i>Proposes problem or issue to be explored; determines purpose of lab</i>	<i>Addresses or plans procedure to be used</i>	<i>Writes purpose and planned procedure in own words</i>	<i>Carries out procedure , records actions, and records data</i>	<i>Supplies answers, results, and/or conclusions</i>	<i>Follow-up: proposal of new ideas &amp; experiments; discussion of broader impact</i>
<b>Level 0</b> Demonstration or minor activity	Teacher	Teacher	Teacher	Teacher or Student	Teacher or Student	Teacher
<b>Level I</b> Data Collection & Analysis	Teacher	Teacher	Teacher	Student	Student	Teacher
<b>Level II</b> Experimental Procedure	Teacher	Teacher	Student	Student	Student	Student
<b>Level III</b> Formal Research Paper	Teacher or Student	Student	Student	Student	Student	Student

<b>Level I</b> Data Collection & Analysis	<b>100% of points</b> (10/10 or 5/5)	<b>60% of points</b> (6/10 or 3/5)	<b>20% of points</b> (2/10 or 1/5)	<b>No points</b> (0/10 or 0/5)
<b>Assessment</b>	<i>Masterful, fully engaged, able to generate new ideas</i>	<i>Proficient, skilled, competent, meeting expectations; engaged</i>	<i>Possessing of some skill, but still practicing; not fully engaged</i>	<i>Not yet proficient or unable to assess; not engaged</i>
<i>Presentation</i>	Lab sheet is clean, not ripped or crumpled. Data and answers to questions are presented neatly and legibly. There are no cross-outs or scribbles on the report	Lab sheet is mostly clean. Handwriting could be neater and/or more legible. There are some cross-outs.	Lab is messy; there are many cross-outs; handwriting is not legible. Lab sheet is torn and/or crumpled.	Unable to assess work; work not turned in
<i>Data / Observations</i>	All data is presented correctly – correct units, significant digits	Some data and/or observations are not complete; some units are missing.	Some data is missing; units are missing	Unable to assess work; work not turned in
<i>Conclusion Questions</i>	Questions are answered completely and correctly. Answers are in student's own words and not simply copied from the book or class notes	Questions are completed, but not correctly; (or) answers lack insight into the material	Not all questions are completed	Unable to assess work; work not turned in

<b>Level II</b> Experimental Procedure	<b>100% of points</b> (10/10 or 5/5)	<b>60% of points</b> (6/10 or 3/5)	<b>20% of points</b> (2/10 or 1/5)	<b>No points</b> (0/10 or 0/5)
<b>Assessment</b>	<i>Masterful, fully engaged, able to generate new ideas</i>	<i>Proficient, skilled, competent, meeting expectations; engaged</i>	<i>Possessing of some skill, but still practicing; not fully engaged</i>	<i>Not yet proficient or unable to assess; not engaged</i>
<b>Presentation &amp; Procedure</b>	Lab sheet is clean and intact. Material is presented neatly and legibly. Student has written the purpose, procedure, data tables, and conclusions in his/her own words. Procedure is written down <i>during the lab</i> , not before or after.	Lab sheet is mostly clean and intact. Handwriting could be neater and/or more legible. Student is missing or has left incomplete one important element.	Lab is messy; handwriting is not legible. Lab sheet is torn and/or crumpled. Student is missing or has left incomplete several important elements.	Unable to assess work; work not turned in
<b>Data &amp; Observations</b>	All data is presented correctly – correct units, significant digits. Data table is clear and makes sense to the reader.	Some data and/or observations are not complete; some units are missing.	Some data is missing; units are missing	Unable to assess work; work not turned in
<b>Results &amp; Conclusions</b>	Results and conclusions are in student's own words. Student has briefly and expertly connected the purpose of the lab to the results and conclusions.	Results and conclusions, though present, lack insight into the material or are not connected to the purpose of the laboratory	Lab report is missing results and/or conclusions	Unable to assess work; work not turned in

Level III Research Paper	A / 4.0	B / 3.0	C / 2.0	D / 1.0 to F / 0.0
<b>Assessment</b>	<i>Masterful, fully engaged, able to generate new ideas</i>	<i>Proficient, skilled, competent, meeting expectations; engaged</i>	<i>Possessing of some skill, but still practicing; not fully engaged</i>	<i>Not yet proficient or unable to assess; not engaged</i>
<i>Written framework</i>	Written work <b>generates a new idea</b> that goes beyond the expectations of the assignment; report is <b>suitable for use as a model</b> for future students; report could be <b>easily read and understood</b> by a person with no access to the lab instructions	Writing <b>meets grade-level standards</b> ; all <b>core components</b> of the report are easy for a fellow student to read and understand; and relate to one another in a logical way; layout is <b>clean and easy to follow</b> ; tables, charts, and graphs are <b>informative</b> and well-designed	Writing may not meet grade-level standards; work may contain <b>substantial errors in communication</b> , including in overall layout, flow, units, significant figures, equations, tables, charts and/or graphs	Incomplete or poorly written framework; student requires <b>review of fundamental concepts</b> and methods; or missing core components; or reader is unable to assess work; or plagiarized
<i>Laboratory procedure</i>	Procedure <b>transcends or exceed expectations</b> in some insightful and unique way; student demonstrates <b>fluency</b> in the scientific method; procedure demonstrates <b>deep understanding</b> of the limitations of the equipment or data set	The work is <b>grounded in the scientific method</b> ; experiment is designed to include <b>adequate controls</b> ; enough <b>trials</b> are performed to reduce error; reader has <b>direct access to the data</b> in question	Work is <b>lacking in one or more aspects</b> of the scientific method; procedure <b>lacks adequate controls</b> or <b>sufficient trials</b> to reduce error; reader cannot <b>fully and independently evaluate</b> the procedure	Work does not demonstrate <b>competency</b> in the scientific method; or the results are not reliable; or reader is unable to assess work; or plagiarized
<i>Analysis</i>	In addressing any apparent difference between the predictions of theory and the experimental results, the student's analysis <b>generates new hypotheses</b> and suggests <b>new experiments</b> ; the report closes some doors while opening others	The results of the experiment <b>clearly support</b> (or <b>clearly fail to support</b> ) the hypothesis; errors are <b>addressed</b> and accounted for; theory and experiment are compared in a <b>logical</b> way; suggestions for <b>improvement</b> of the experiment are offered; analysis of error is taken seriously	<b>Some care</b> was taken to account for and reduce errors; however, review and analysis of errors during the experiment should have led the student to experimental <b>redesign</b> or <b>repetition</b> ; analysis of error is treated as an afterthought	Analysis of experimental results and errors is <b>incomplete</b> or <b>missing</b> ; or the experimental results are not <b>clearly related</b> to the hypothesis; or reader is unable to assess work; or plagiarized
<i>Other expectations</i>	Student goes beyond the expectations of the teacher and course by <b>broadening the scope</b> of the experiment to include novel, <b>scientifically valid</b> hypotheses, procedures, and analyses	A <b>quantitative</b> analysis of experimental error is made; graphs are <b>linearized</b> when appropriate and include <b>best-fit lines</b> ; student demonstrates competency in the use of <b>graphical analysis software</b> ; probes and sensors are <b>properly calibrated</b> ; student is heavily involved in the <b>design</b> of the experiment itself	Analysis of experimental error lacks a quantitative component; analysis is not grade-level appropriate; improper or missing calibration and testing of equipment; student is not <b>involved</b> in experimental design	Level-specific expectations <b>have not been met</b> ; or, reader is unable to assess work; or plagiarized

Level III Research Paper	The Scientific Method – a guide for SI students
<b>What is a scientific theory?</b>	<p>A scientific theory is an explanation of how the natural world works. A theory must have <i>predictive power</i>, meaning that it can tell you what will happen next in a given situation. A theory can be considered <i>credible</i>, but must always be understood to be <i>conditional</i> (never “proven”). In other words, we must always be open to the idea that a theory will later be replaced by a more accurate theory that explains the world even better. Key theories include: that, over long periods of time, random mutations in the genetic code of a species can lead to beneficial changes that better allow the species to survive and prosper in its environment (<i>biology</i>); that the organization of the electrons in an oxygen atom make it a volatile substance – when oxygen is combined with hydrogen gas and a spark, an explosion will occur that produces pure water (<i>chemistry</i>); and that the apparent force of gravity can be explained as a curvature in the fabric of space-time – near massive objects, both space and time are stretched and distorted in measurable ways (<i>physics</i>). The scientific method is grounded in the notion of cause-and-effect – just because two things happened together doesn’t mean one caused the other; both could be due to some other, unknown underlying cause.</p>
<b>How is a scientific theory tested?</b>	<p>A scientist makes careful experimental observations to confirm a certain prediction of a theory. The process typically begins with the formation of a hypothesis. Several things are required: the experiments must be <i>controlled</i> to make sure the hypothesis is being <i>uniquely</i> tested. For example, if it is my hypothesis that a certain gene leads to breast cancer, I might study the lives of identical twins separated at birth so that I can distinguish hereditary effects from environmental effects. A theory is said to be <i>confirmed</i> if the predictions made by the theory come true under experiment. A theory is said to be <i>falsified</i> if the predictions do not come true. Sometimes a theory that has been confirmed under a certain set of conditions but falsified under other conditions, is still useful. (For instance, we still use Newton’s theory of gravity when calculating simple spacecraft trajectories even though we know the theory is false in certain extreme conditions.)</p>
<b>When should we amend a theory?</b>	<p>Discrepancies between theory and experiment can be <i>real</i> (meaning that the theory is wrong, and must be amended), or can be due to <i>experimental error</i>. Experimental error can be separated into two types: statistical and systematic. <i>Statistical errors</i> arise when too few measurements are made to fully represent the range of possible experimental outcomes. (For instance, if you poll 100 random students about how they feel about the dress code, your results will be less accurate than if you polled 500 random students.) <i>Systematic errors</i> are related to the accuracy of measurement. (For example, if you are using a balance to measure the weight of a liquid sample but the balance reads 5 g high due to a problem with its calibration.) When a scientist discovers a discrepancy between theory and experiment that cannot be explained by experimental error, a new discovery has been made and the theory must be amended. Scientists constantly strive to reduce all error so the comparison between theory and experiment can be accurately made. All types of error must be qualitatively and quantitatively accounted for and understood – vague references to unknown and uncorrected mistakes are not scientific.</p>
<b>How do we share our scientific results?</b>	<p>Research papers include the following <i>core components</i>: an <b>abstract</b> (a brief, to-the-point summary of the entire report), an <b>introduction</b> (detailing the theory &amp; hypothesis), a discussion of <b>experimental procedure</b> (explaining how the experiment is performed in detail), a presentation of the <b>experimental results</b> (often, but not always, presented in data tables or graphs), an <b>analysis</b> of the results (including but not limited to a discussion of experimental error), and a series of <b>conclusions</b> (summarizing the report). Scientists reading a report expect your methods to be clear, truthful, and reliable. Others should be able to understand and repeat your work just based on your descriptions, and they should expect to arrive at the same conclusions you did. There are no hidden data or secret techniques. Everything is out in the open. This transparency generates honesty, accountability, and accuracy in the scientific community. All observations must be clear and truthful, and each measurement must be accompanied by estimates of the corresponding experimental errors. (<i>Note</i>: not all scientific endeavors require formal lab reports. At times at SI you’ll be asked to complete worksheets, informal lab reports, work on “lab practical” exams, participate in challenges, and the like.)</p>

## Level III Research Paper

### Statement of Problem

*Example:* What is the effect of exposure to UV radiation on cancer rates?

- Not a yes/no question
- Statement narrows down a topic area
- Generalized variables included
- Problem is clearly testable

### Hypothesis

*Example:* We predict increased cancer rates as a result of longer exposure to UV radiation, which can cause mutations in DNA molecules.

- Statement predicts a relationship or trend
- Statement gives specific direction to the prediction: a stand is taken
- Prediction includes both independent and dependent variables
- A rationale is given for the hypothesis

### Materials and Procedure

*Example:* Lists of materials; well-organized procedure that can be followed by a novice reader; useful diagrams and sketches

- All materials used are listed properly
- Materials are listed separately from procedure
- Procedure is well-organized, in logical sequence
- Enough information is given so another student could repeat the procedure
- Helpful diagrams and sketches are used
- Repeated trials are performed

### Experimental Controls

*Example:* Our experimental subjects will be of the same age, all non-smokers, and of roughly equal socioeconomic status.

- Number of controlled variables identified: \_\_\_\_\_
- Experimental controls correctly identified
- Experimental controls make logical sense

### Qualitative Observations

*Example:* When we heated substance A with a flame, thick white smoke and charred ash was produced

- Observations are made during the procedure
- Deviations from expectations are noted
- Observations and discussion of results are provided

### Variables

*Example:* The independent variable is exposure time; the dependent variable is cancer rate.

- Independent variable correctly identified
- Independent variable correctly defined
- Dependent variable correctly identified
- Dependent variable correctly defined

### Quantitative Data

*Example:* Data tables and data graphs

- All raw data is provided
- All data is provided with appropriate units
- All data is reported with correct significant figures
- A condensed table with the most important data is included
- Tables are properly labeled and include header rows
- Example calculations are provided
- Appropriate graph types are used
- Graphs are titled
- Graphs are labeled properly (axes & series)
- Units are included in all graphs
- Trends in data are represented by best-fit curves
- Appropriate scales are used in graphs

### Statistics

*Example:* Measured accelerations ranged from 0.51 m/s<sup>2</sup> to 0.82 m/s<sup>2</sup> with an average of 0.66 m/s<sup>2</sup>

- Measure of central tendency (average / mean)
- Measure of variation (range / deviation)
- Regression analysis (best-fit line)
- Other appropriate statistic used

### Analysis and Interpretation

*Example:* Among those with 4 or more hours of directly sunlight per day, the average cancer rate was 20.5 per 10,000.

- All data presented and discussed
- All data interpreted for meaning
- Unusual data points pointed out
- Unusual data points explained
- Trends in data pointed out
- Trends in data explained
- Enough detail provided to understand data
- Response is clear and concise
- All statements are supported by the data

### Experimental Error

*Example:* Since we surveyed 100 people, our statistical error is  $1 / \sqrt{100} = 0.1 = 10\%$ .

- Systematic errors are discussed
- Statistical (random) errors are discussed
- Human mistakes are addressed and/or corrected during the experiment
- The effect of data errors is discussed
- Errors are quantified

### Conclusion

*Example:* Our results confirmed our hypothesis: those with higher exposure levels to UV radiation were indeed more likely to develop cancer.

- Hypothesis is evaluated according to data
- Hypothesis is re-stated
- Reasons to accept/reject the hypothesis are given
- All statements are supported by the data

### Recommendations

*Example:* Future improvements to this experiment would include ...

- Suggestions for improvement of experiment are given
- Suggestions for future experiments are given